

Final Report on the Safety Assessment of Glycol Stearate, Glycol Stearate SE, and Glycol Distearate

Glycol Stearate, Glycol Stearate SE, and Glycol Distearate consist primarily of the mono- and diesters of triple-pressed stearic acid. They are used in numerous categories of cosmetic products at concentrations ranging from less than 0.1 to 10%.

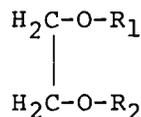
Animal data for acute oral toxicity, skin and eye irritation, and sensitization show that these ingredients have low acute toxicity. A repeated insult patch test with 50% Glycol Distearate on 125 subjects presented no evidence of skin irritation or hypersensitivity. Human studies using formulations containing Glycol Stearate at levels of 2-5% reported no skin irritation or sensitization.

Subchronic testing has not been adequately investigated in laboratory animals. Human test data for formulations containing > 4% Glycol Stearate or Glycol Distearate should be considered.

Based on the available information presented herein, it is concluded that Glycol Stearate, Glycol Stearate SE, and Glycol Distearate are safe as cosmetic ingredients in the present practices of use and concentration.

CHEMICAL AND PHYSICAL PROPERTIES

These ingredients are mixed esters of ethylene glycol and triple-pressed stearic acid. The latter consists of 42.5% stearic acid and about an equal amount of palmitic acid, along with lesser amounts of several other fatty acids. The general structural formula for these ingredients is:^(1,2)



Glycol Stearate: The ingredient is comprised of 40-70% of the monoester in which R₁ is the acyl portion of triple-pressed stearic acid and R₂ is H. Glycol

Stearate also contains a significant portion, 30–58%, of the diester in which both R_1 and R_2 are the acyl moiety of triple-pressed stearic acid.⁽²⁾

Glycol Stearate SE: This ingredient is a self-emulsifying grade of Glycol Stearate containing free stearic acid and some sodium and/or potassium stearate.⁽¹⁾

Glycol Distearate: This ingredient is the diester of ethylene glycol in which both R_1 and R_2 are the acyl moiety of triple-pressed stearic acid.⁽²⁾

Glycol Stearate, Glycol Stearate SE, and Glycol Distearate have similar physical properties. They are white to cream colored waxy solids. Their physical properties vary within specified limits according to their proportions of mono- and diesters and other components. Depending on the intended use, a purchasing specification is used to set specific limits on the physical characteristics of these ingredients.⁽²⁾

Analytical Methods

Glycol Stearate and Glycol Distearate can be analyzed by gas chromatography.⁽³⁾ Mass spectrometric analysis of long-chain esters of ethanediol (ethylene glycol) has been described⁽⁴⁾; this allows for the identification of individual esters of the diol as well as of classes of diol monoesters. A method of gel-permeation chromatography for Glycol Distearate on Sephadex LH-20 has also been reported.⁽⁵⁾ Standard methods have been suggested for determining the chemical properties of these ingredients.⁽²⁾

Impurities

Impurities such as free stearic acid (triple-pressed), the mono- or diesters, ethylene glycol, and corresponding derivatives of other fatty acids found in the stearic acid may be present in Glycol Stearate.⁽²⁾

Ethylene glycol and/or ethylene oxide are used as starting material for the synthesis of Glycol Stearate. Since the former is known to be contaminated with traces of 1,4-dioxane,⁽⁶⁾ it is possible that such traces also appear in the synthesized material. Analytical data on traces of 1,4-dioxane in Glycol Stearate were not available to the Expert Panel.

When rats were given high doses of 1,4-dioxane in drinking water (~1.0%) for 13 months, liver lesions including hepatomas occurred.⁽⁷⁾

USE

Purpose and Frequency of Use in Cosmetics

These ingredients are used as emulsifiers, dispersants, opacifiers, and viscosity modifiers. As wax ingredients in stick preparations, they have served to control hardness, add slip, and increase opacity. They give lotion, cream, and detergent formulations an opaque or milky appearance.^(8,9)

As shown in Table 1, these ingredients are used in a variety of categories of cosmetic products; their concentrations range from less than 0.1% to as high as 10%. The cosmetic product formulation computer printout which is made

ASSESSMENT: GLYCC

TABLE 1. Proc

Cosmetic pro
Ingred.

Glycol Stearate
Bath oils, tablets
Bubble baths

Other bath prep
Eyebrow pencil
Eyeliner
Eyeshadow

Mascara
Hair conditione
Hair straightene
Rinses (noncolo
Shampoos (non

Tonics, dressing
hair grooming
Hair shampoos
Blushers (all typ
Foundations
Lipsticks
Makeup bases
Rouges
Other makeup p
Bath soaps and

Aftershave lotio
Cleansing (cold
cleansing lotic
and pads)
Face, body and
(excluding sha
preparations)
Moisturizing

Other skin care

Suntan gels, cre
liquids

Glycol Stearate
Other skin care

Glycol Distearat
Hair conditione
Permanent wavi
Shampoos (non

Hair dyes and c
types requirin
statement and

diester in which both (2)

ing grade of Glycol and/or potassium

ylene glycol in which acid.⁽²⁾

tearate have similar solids. Their physical proportions of monoended use, a purchased characteristics of

d by gas chromatography ethanediol (ethylene mon of individual esters od of gel-permeation H-20 has also been for determining the

re mono- or diesters, tity acids found in the

arting material for the be contaminated with pear in the synthesized col Stearate were not

inking water (~ 1.0%) d.⁽⁷⁾

etics

pacifiers, and viscosity have served to control , cream, and detergent

variety of categories of han 0.1% to as high as ntout which is made

TABLE 1. Product Formulation Data.^a

Cosmetic product type/ Ingredient	Concentration (%)	No. of product formulations
<i>Glycol Stearate</i>		
Bath oils, tablets and salts	>0.1-1	6
Bubble baths	>1-5	3
	>0.1-1	44
Other bath preparations	>0.1-1	6
Eyebrow pencil	>1-5	3
Eyeliners	>1-5	9
Eyeshadow	>5-10	1
	>1-5	75
Mascara	>1-5	2
Hair conditioners	>5-10	2
Hair straighteners	>5-10	4
Rinses (noncoloring)	>0.1-1	3
Shampoos (noncoloring)	>5-10	1
	>1-5	46
	>0.1-1	28
	≤0.1	2
Tonics, dressings, and other hair grooming aids	>1-5	1
Hair shampoos (coloring)	>1-5	2
Blushers (all types)	>1-5	5
Foundations	>1-5	88
Lipsticks	>1-5	1
Makeup bases	>1-5	2
Rouges	>1-5	8
Other makeup preparations	>1-5	2
Bath soaps and detergents	>1-5	1
	>0.1-1	1
Aftershave lotions	>0.1-1	1
Cleansing (cold creams, cleansing lotions, liquids, and pads)	>1-5	3
	>0.1-1	5
Face, body and hand (excluding shaving preparations)	>1-5	9
	>0.1-1	2
Moisturizing	>5-10	1
	>1-5	8
	>0.1-1	3
Other skin care preparations	>5-10	2
	>1-5	2
	>0.1-1	1
Suntan gels, creams, and liquids	>1-5	1
<i>Glycol Stearate SE</i>		
Other skin care preparations	>0.1-1	1
<i>Glycol Distearate</i>		
Hair conditioners	>0.1-1	1
Permanent waves	>1-5	5
Shampoos (noncoloring)	>1-5	9
	>0.1-1	6
Hair dyes and colors (all types requiring caution statement and patch test)	>0.1-1	1

TABLE 1. (Continued.)

Cosmetic product type/ Ingredient	Concentration (%)	No. of product formulations
Deodorants (underarm)	> 1-5	1
Other personal cleanliness products	> 5-10	1
Other shaving preparation products	> 1-5	1
Cleansing (cold creams, cleansing lotions, liquids, and pads)	> 1-5	1

^aData from Ref. 10.

available by the Food and Drug Administration (FDA) is compiled through voluntary filing of such data in accordance with Title 21 part 720.4 of the Code of Federal Regulations (1979). Ingredients are listed in prescribed concentration ranges under specific product type categories. Since certain cosmetic ingredients are supplied by the manufacturer at less than 100% concentration, the value reported by the cosmetic formulator may not necessarily reflect the true, effective concentration found in the finished product; the effective concentration in such a case would be a fraction of that reported to the FDA. The fact that data are only submitted within the framework of preset concentration ranges also provides the opportunity for overestimation of the actual concentration of an ingredient in a particular product. An entry at the lowest end of a concentration range is considered the same as one entered at the highest end of that range, thus introducing the possibility of a two- to ten-fold error in the assumed ingredient concentration. According to FDA, Glycol Stearate SE is used in one unspecified skin-care product. Glycol Distearate is principally employed in hair-care preparations⁽¹⁰⁾; however, its use as a lyophilic component of self-emulsifying ointment bases has been described.⁽¹¹⁾

Products containing these ingredients are used on all body orifices. Thus they may enter the body by several routes (though the inhalation of sprays appears to be minor as a mode of exposure and absorption).

These ingredients may be applied as often as several times a day (lipsticks and lotions) or as infrequently as once every one or two months (hair dyes and colors). The period of time for which they remain in contact may be conditioned by the frequency with which the affected part of the body is washed.

BIOLOGICAL PROPERTIES

General Effects

The addition of 12.5 percent Glycol Stearate as a surfactant to a vaseline-based ointment increased the cutaneous absorption of the following compounds through the shaved skin of rats by the factors shown: 10% potassium iodide (4X); 5% sodium salicylate (4.6X); and 5% ammonium thiocyanate (3.1X). A two-gram sample of each emulsion was rubbed into the skin for five minutes and then covered with a protective bandage. Absorption was determined by the analysis of urine specimens collected at 12 and 24 hours.⁽¹²⁾

ASSESSMENT: GLYCC

Oral Toxicity:

in five studies for marized in Table weight in corn oil nasal hemorrhag ministration, but with high levels of the 14-day gross a the test material.⁽⁶⁾

For 91 days, f five females, were dients was ethyle and 5%. The equ 0.0025-0.0125%, histopathologic e and test groups.⁽⁶⁾

Primary Skin

Glycol Stearate, C albino rabbit skin ritating (See Tabl were tested for cc Transportation, tl

Sensitization

Stearate and Gly the shaven back ml injection, 0.1 jections. Two w taken 24 hours l

Subchronic:

mulation was ap female rats. Obs behavior, hema histopathologic c of the test formu

A shampoo separate experir three females). / male and five fer per week to inta this remained or

Two formul: Distearate applic treatment-induce was reported to

Animal Toxicology

Oral Toxicity: Glycol Stearate and Glycol Distearate have each been tested in five studies for acute oral toxicity in rats; the data from these studies are summarized in Table 2. During the various studies, doses of 13 or more g/kg body weight in corn oil produced effects which included diarrhea, wet oily coats, and nasal hemorrhage; the symptoms appeared within four days following administration, but disappeared within the next six days. No animals were dosed with high levels of corn oil alone. One study on Glycol Distearate reported that at the 14-day gross autopsy, the stomach contained residues which appeared to be the test material.⁽¹³⁾

For 91 days, four groups of weanling rats, each comprised of five males and five females, were fed a diet containing a dishwashing liquid one of whose ingredients was ethylene glycol distearate at a concentration range of between 1% and 5%. The equivalent dosing levels of the ethylene glycol distearate were 0, 0.0025–0.0125%, 0.005–0.025%, and 0.01–0.05%. Following both gross and histopathologic examination, no differences were observed between the controls and test groups.⁽¹⁴⁾

Primary Skin Irritation Studies: Draize type procedures were used to test Glycol Stearate, Glycol Stearate SE, and Glycol Distearate for primary irritation of albino rabbit skin; the ingredients were found to be nonirritating to slightly irritating (See Table 2). In addition, when Glycol Stearate and Glycol Distearate were tested for corrosivity according to the procedures of the U.S. Department of Transportation, they were found to be noncorrosive to rabbit skin.⁽¹³⁾

Sensitization: Sensitization studies were conducted in guinea pigs on Glycol Stearate and Glycol Distearate. Each ingredient was injected intradermally into the shaven back of each of two male, white guinea pigs. Following an initial 0.05 ml injection, 0.1 ml injections were given three times a week for a total of ten injections. Two weeks later a challenge injection was given, and readings were taken 24 hours later. Both ingredients were found to be nonsensitizing.⁽¹³⁾

Subchronic: For 90 days, Glycol Stearate at 3% in a liquid foundation formulation was applied five times a week for 13 weeks to the clipped backs of 15 female rats. Observations were made for survival, body weight, appearance and behavior, hematology, clinical chemistry, organ weights, and gross and histopathologic changes. No effects were attributed to the repeated application of the test formulation.⁽¹³⁾

A shampoo formulation containing Glycol Distearate was tested in three separate experiments on groups containing six rabbits each (three males and three females). A fourth experiment involved similar procedures, but had five male and five female rabbits per group. The material was applied daily, five days per week to intact or abraded skin equivalent to 10% of the skin area of the back; this remained on the animal for seven hours each day before washing.⁽¹⁴⁾

Two formulations were tested for 91 days. The concentration of Glycol Distearate applied to the animals ranged from 0.05% to 0.5%. No evidence of treatment-induced systemic effects was observed. The skin irritation that resulted was reported to be similar to that produced by other forms of shampoo.⁽¹⁴⁾

No. of product formulations
1
1
1
1

mpiled through volun-
720.4 of the Code of
scribed concentration
1 cosmetic ingredients
centration, the value
flect the true, effective
concentration in such
fact that data are only
nges also provides the
1 of an ingredient in a
ntration range is con-
ange, thus introducing
redient concentration.
ecified skin-care prod-
-care preparations⁽¹⁰⁾;
ng ointment bases has

ill body orifices. Thus
halation of sprays ap-
).

nes a day (lipsticks and
months (hair dyes and
ct may be conditioned
ly is washed.

urfactant to a vaseline-
: following compounds
potassium iodide (4X);
ate (3.1X). A two-gram
five minutes and then
rmined by the analysis

Table 2. Acute Animal Toxicity.^a

Ingredient	File No.	LD50—Acute oral			No. Rats/ Dose	Skin Irritation		Animals	Draize Woodward Calvary Irritation Index	Eye Irritation	Draize No. of Rabbits	Comment
		Value	Conc.	Dosage		Conc.	Conc.					
Glycol Stearate	6.4b.i	>10 g/kg	50% in corn oil	0.464–10 g/kg	5	undiluted	6 rabbits	0.13	undiluted	6	mild transient irritant in 1/6	
	6.4b.ii	>21.3 g/kg	1:2 in corn oil	0.7–21.3 g/kg	5	undiluted	6 rabbits	0.0	undiluted	9	no irritation	
	6.4b.ii	[Dept. of Transportation >10 g/kg	(Dept. of Transportation Skin Irritation Test)	undiluted	10	undiluted	6 rabbits	0.0	undiluted	6 rabbits	0.0	
	6.4b.iv	>10 g/kg	undiluted	10 g/kg	10	undiluted	6 rabbits	0.375	undiluted	6 rabbits	0.375	
6.4b.iv	[Skin Sensitization Test]					i.c. inject. of 0.1% in saline	2 guinea pigs	not a sensitizer				
Glycol Stearate SE	6.4d.i	>5000 mg/kg	undiluted	5000 mg/kg	10	undiluted	3 rabbits	0.8	undiluted	3	practically non- irritating	
Glycol Distearate	6.4c.i	>10 g/kg	50% in corn oil	0.464–10 g/kg	5	undiluted	6 rabbits	0.04	undiluted	6	practically non- irritating	
	6.4c.ii	>16 g/kg	1:4 in corn oil	0.5–16 g/kg	5	5% in water	3 rabbits	0.0	5% in water	3	not an irritant	
6.4c.ii	[Dept. of Transportation >10 g/kg	(Dept. of Transportation Skin Irritation Test)	undiluted	10	undiluted	6 rabbits	6 rabbits	0.0	undiluted	6	practically non- irritating	
6.4c.iv	>10 g/kg	undiluted	10 g/kg	10	undiluted	6 rabbits	6 rabbits	0.085	undiluted	9	no irritation	
6.4c.iv	[Skin Sensitization Test]					i.c. inject. of 0.1% in saline	2 guinea pigs	not a sensitizer				
4.4c.v	>5000 mg/kg	undiluted	5000 mg/kg	10	undiluted	3 rabbits	3 rabbits	1.0	undiluted	3	practically non- irritating	

^aData from Ref. 14.

Two formulations of Glycol Distearate ranged from 19 to 76 years of age. Microscopic examination of the skin showed no systemic toxic effects by the surfactant.

A separate but unpublished study on Glycol Distearate associated with both formulations reported noted no "toxicity" to the test.⁽¹⁴⁾

A shampoo containing concentrations of 0.05% and 0.1% concentration. After 14 days from the application, no irritation was observed in one formulation.⁽¹⁴⁾

Eye Irritation
These three ingredients were found to be nonirritating to the eyes.

Potential Toxicity
Glycol Stearate, the toxic component, is hydrolyzed by skin moisture to release glycerol. A review of the literature indicates that it has a low potential for toxicity which might be expected.

Unpublished data reviewed and are available.

Skin Irritation
Glycol Distearate was applied from 19 to 76 years of age to the dorsum of the back for a 24-hour induction period. Challenge patches were applied for a final insult patch; no irritation was observed in any of the formulations.

Eyeshadow
Glycol Distearate was applied sequentially to the eyelids (Glycol Distearate), blusher

r to the start of the test. idy began and at one- iod. The dermatologist n over the entire four- ts tested demonstrated (16)

lay cumulative irritancy eliner containing 3.5% occlusive patch. A maximum-effect basis was mum group value was ; 0.08. (16)

: The formulation was t subjects. The average possible score of 84.0. ts, a score of 47.5 was gle individual. (16)

A repeated insult patch nsidered sensitive. The id not evoke any reac- ; were stated, and the

cream containing 2.5% tch test procedure was indication of sensitiza- of the repeated patch

rate: A repeated insult dnesday, and Friday of quid solution of the for- r arm of each subject. he diluted test material tearate.) Fourteen days was challenged with a nined 48 and 96 hours d. (16)

ite: A repeated insult of a 0.2% solution of a 2–5% ethylene glycol ontained 0.004–0.01% entical to those in the ed. (16)

glycol distearate were a three-week period, te days. Fourteen days ven challenge patches.

TABLE 3. Sensitization Tests on Dishwashing Liquids Containing Ethylene Glycol Distearate.^a

Dishwashing liquid	No. of subjects	Detergent conc. (%)	Range of conc. of ethylene glycol distearate (%)
1	67	1	0.01–0.05
2	69	1	0.01–0.05
3	87	1.5	0.015–0.075
4	78	0.5	0.005–0.025

^aData from Ref. 14.

Table 3 shows the range of concentration of ethylene glycol distearate for each group of subjects.

No results were presented on irritation caused by the test compounds. In all cases, there was no reported evidence of sensitization after challenge.⁽¹⁴⁾

Consumer Information: Two companies reported on the incidence of consumer complaints related to their products containing Glycol Stearate. One indicated that it was unaware of any complaints having arisen over a 20-year period from the use of over two million units of products (various creams and lotions) containing 0.5–5% Glycol Stearate. According to the second company, the unscreened adverse reaction rate for shampoos containing 4.0% Glycol Stearate averaged 1.2 complaints per million.⁽¹⁴⁾

Occupational Exposure: Two manufacturers reported that they have been manufacturing Glycol Stearates and Glycol Distearates for between 20 and 30 years. According to both, no employee reported that his or her health might have been adversely affected by exposure to these compounds. This conclusion was based upon: (a) 30 employees who for 10 years had potentially been exposed to Glycol Stearate for 1% of their work time; (b) 70 employees who for 20 years had potentially been exposed to Glycol Distearate for 20% of their work time; and (c) 50 employees who for 30 years had potentially been exposed to Glycol Stearate for 5% of their work time. One manufacturer noted that its labor turnover was very low, so that some individuals had been exposed to the ingredients for many of the years during which they had been produced there.⁽¹⁴⁾

SUMMARY

Glycol Stearate, Glycol Stearate SE, and Glycol Distearate are comprised primarily of the mono- and diesters of triple-pressed stearic acid. They are used at concentrations ranging from less than 0.1% to 10% in numerous categories of cosmetic products. They function as emulsifiers, dispersants, opacifiers, and viscosity modifiers, and have been used as wax ingredients in stick preparations. Because they are used on all body surfaces, these ingredients may be absorbed through several routes; and their contact with the body may be frequent and prolonged. Animal studies indicate that Glycol Stearate serves as a surfactant and enhances percutaneous absorption.

The animal data indicate that these ingredients have low acute oral toxicity, skin and eye irritation, and sensitization. One subchronic skin painting study with a product formulation containing 3% Glycol Stearate showed no toxic effects throughout the 90-day test period and after necropsy.

A repeated insult patch test with 50% Glycol Distearate on 125 subjects presented no evidence of skin irritation or hypersensitivity. Human studies using formulations containing Glycol Stearate at levels of 2–5% reported no skin irritation or sensitization. Additional human studies using Glycol Distearate, at levels of the test compound 500 times lower than that which a consumer would actually use, showed no irritation or sensitization upon challenge. Prolonged repeated insult patch testing on the forearm was used to approximate the high-level exposure consumers would experience when they applied a shampoo containing Glycol Distearate to their scalps, under hot and wet conditions, for a very short period of time.

Subchronic testing has not been adequately investigated in laboratory animals. Human test data for formulations containing > 4% Glycol Stearate or Glycol Distearate should be considered.

CONCLUSION

On the basis of the available information presented herein, the Panel concludes that Glycol Stearate, Glycol Stearate SE, and Glycol Distearate are safe as cosmetic ingredients in the present practices of use and concentration.

REFERENCES

1. ESTRIN, N.F. (ed.). (1977). *CTFA Cosmetic Ingredient Dictionary*, 2nd ed. Washington, DC. Cosmetic, Toiletry and Fragrance Association.
2. COSMETIC, TOILETRY AND FRAGRANCE ASSOCIATION (CTFA). (1978) Submission of data by CTFA. Cosmetic ingredient descriptions for glycol stearate, glycol stearate SE, and glycol distearate.*
3. SUFFIS, R., SULLIVAN, T.J., and HENDERSON, W.S. (1965). Identification of surface-active agents as trimethyl silyl ether derivatives by gas chromatography. *J. Soc. Cosmet. Chem.* **16**(13), 783–94.
4. BAUMANN, W.J., SEUFERT, J. HAYES, H.W., and HOLMAN, R.T. (1969). Mass-spectrometric analysis of long-chain esters of diols. *J. Lipid Res.* **10**(6), 703–9.
5. CALDERON, M. and BAUMANN, W.J. (1970). Naturally occurring diol lipids. VI. Fractionation of neutral lipids on a lipophilic dextran gel. *Biochem. Biophys. Acta.* **210**(1), 7–14.
6. BIRKEL, T.J., WARNER, C.R., and FAZIO, T. (1979). Gas chromatographic determination of 1,4-dioxane in polysorbate 60 and polysorbate 80. *J. Assoc. Offic. Anal. Chem.* **62**, 931–36.
7. ARGUS, M.F., SOHAL, R.S., BRYANT, G.M., HOCH-LIGETI, C., and ARCOS, J.C. (1973). Dose response and ultrastructural alterations in dioxane carcinogenesis: inference of methylcholanthrene on acute toxicity. *Europ. Cancer J.* **9**, 237–43.
8. GLEASON, M.N., GOSSELIN, R.E., HODGE, H.C., and SMITH, R.P. (1969). *Clinical Toxicology of Commercial Products*, 3rd ed. Williams & Wilkins. Baltimore.
9. BALSAM, M.S. and SAGARIN, E. (eds.). (1974). *Cosmetics: Science and Technology*. 2nd ed., 3 vols. John Wiley and Sons. New York.
10. FDA. (Aug. 31, 1976). Cosmetic product formulation data. Food and Drug Administration. Washington, DC.

ASSESSMENT: GLYCOL S

11. PETKOV, L. (1971). TLC Soap Perfum. Cosmet. **4**
12. FEBVRE, R. and ROBLET absorption of potassium 759–65.
13. CTFA. (1979). Submissic and glycol stearate SE.*
14. CTFA. (Aug. 12, 1979). S distearate.*
15. CTFA. (1979). Submissic glycol distearate.*
16. CTFA. (1979). Submissi distearate.*

*Available upon request: Administrator, Cosmetic Ingredient Review, Suite 810, 1110 Vermont Ave., N.W., Washington, DC 20005.

low acute oral toxicity, nic skin painting study ite showed no toxic ef- sy.

earate on 125 subjects y. Human studies using reported no skin irrita- col Distearate, at levels onsumer would actually Prolonged repeated in- nate the high-level ex- a shampoo containing ditions, for a very short

stigated in laboratory 4% Glycol Stearate or

herein, the Panel con- ol Distearate are safe as concentration.

d. Washington, DC. Cosmetic,

) Submission of data by CTFA. d glycol distearate.*

ion of surface-active agents as hem. 16(13), 783-94.

Mass-spectrometric analysis of

ids. VI. Fractionation of neutral

determination of 1,4-dioxane in 36.

OS, J.C. (1973). Dose response Icholanthrene on acute toxicity.

3). Clinical Toxicology of Com-

chnology. 2nd ed., 3 vols. John

g Administration. Washington,

11. PETKOV, L. (1971). TLC Identification of the hydrophilic components of self-emulsifying emulsion bases. Soap Perfum. Cosmet. **44**, 481-83.
12. FEBVRE, R. and ROBLET, M.A. (1963). The influence of several surfactants in ointments on the cutaneous absorption of potassium iodide, sodium salicylate, and ammonium thiocyanate. Ann. Pharm. Franc. **21**(11), 759-65.
13. CTFA. (1979). Submission of data by CTFA. Unpublished safety data on glycol stearate, glycol distearate, and glycol stearate SE.*
14. CTFA. (Aug. 12, 1979). Submission of data by CTFA. Unpublished safety data on glycol stearate and glycol distearate.*
15. CTFA. (1979). Submission of daya by CTFA. Unpublished safety data on human sensitization test with glycol distearate.*
16. CTFA. (1979). Submission of data by CTFA. Unpublished safety data on glycol stearate and glycol distearate.*